

O₂n-Site Oxygen Production

A stylized atomic model featuring two red spherical nuclei. The left nucleus is surrounded by three blue electrons on light blue elliptical orbits. The right nucleus is surrounded by six orange electrons on light orange elliptical orbits. The orbits are arranged in a way that they appear to be interacting or overlapping.

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Outline

- Project Goal
- Brief Theory
- Progression of Project Design
- Design Conclusions
- Business and Economic Analysis

Problem Statement

- Develop a marketable oxygen generator for local onsite production in medical facilities
- This system should compete with current distribution prices

Recommendation

- Two adsorption system, incorporating both N₂ and Argon pressure swing adsorption, is the recommended system
- Onsite cryogenic distillation is not profitable

What We Need

- Hospital Need - Oxygen
 - 3000 liquid gallons per month
(relatively small)
 - 1.24 lb-mol/hr
 - 99.2% Purity- FDA Standards
 - Dry
 - Remove impurities



Process Selection

- **Criteria**
 - Safety: NFPA 50 and NFPA 99
 - Purity: USP Standards
 - Space of system
 - *Cost of Equipment and Operations*

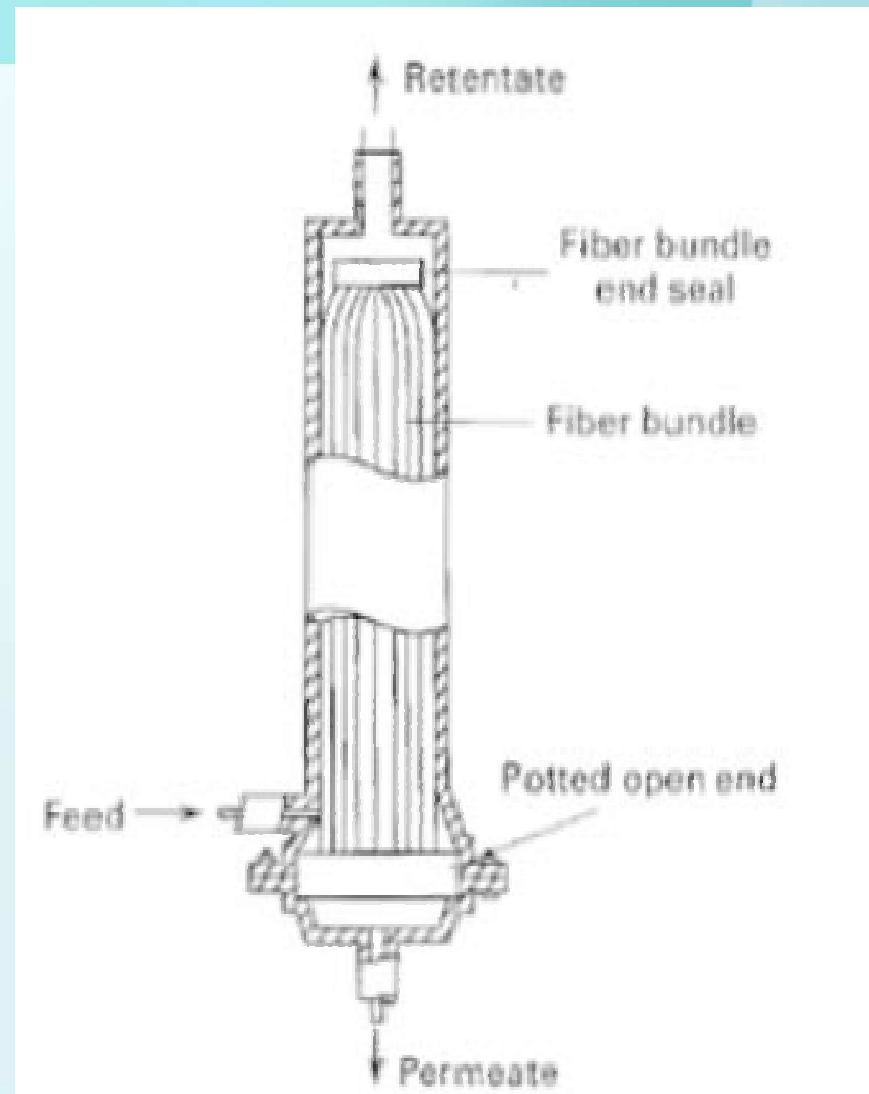
Optimization

- Criteria for optimization
 - Needs of hospital i.e. supply and storage
 - Low maintenance/high convenience
 - Process location and space availability
 - Economics
- Tools for optimization
 - Pro/II
 - Microsoft Excel



Options

- Membrane
 - High purity; still does not achieve needed purity

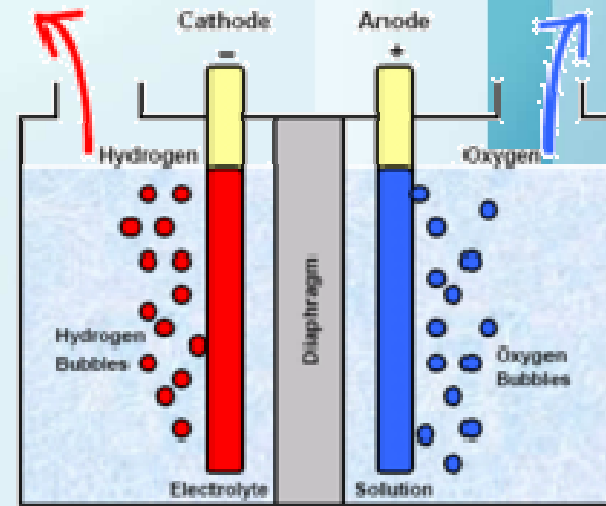


Hollow fiber membrane

Options

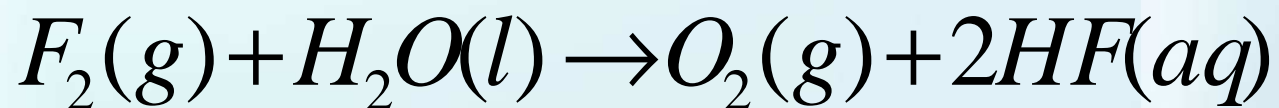
- Electrolysis
 - Process cost is expensive; electricity cost alone is more than twice the cost of buying

- Gibbs Free Energy
- $\Delta G = \Delta H - T\Delta S$
- 450 kJ/mol O₂
→ \$38,000/yr energy costs vs. \$19,000



Options

- Chemical
 - Utilization of a chemical reaction; unwanted product waste



Options

- *Liquefaction*
 - *Can be used to achieve purity of 99.2%*
- *Pressure swing adsorption*

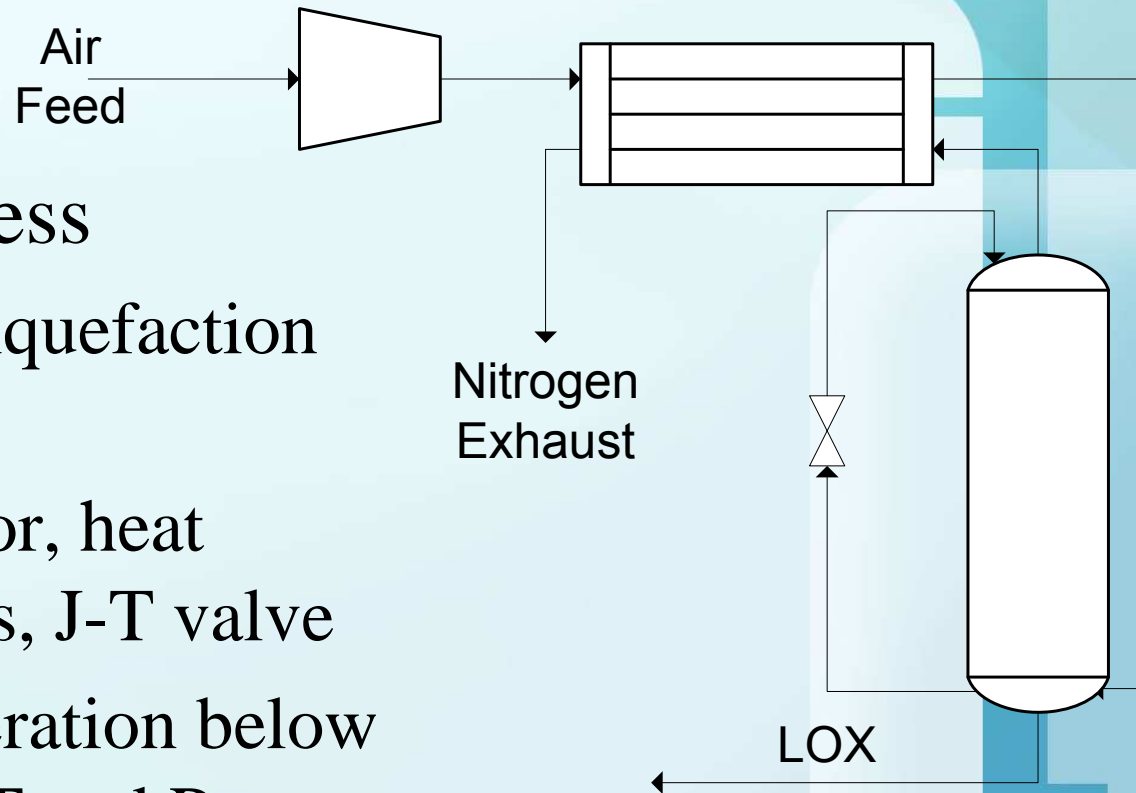
General Theory

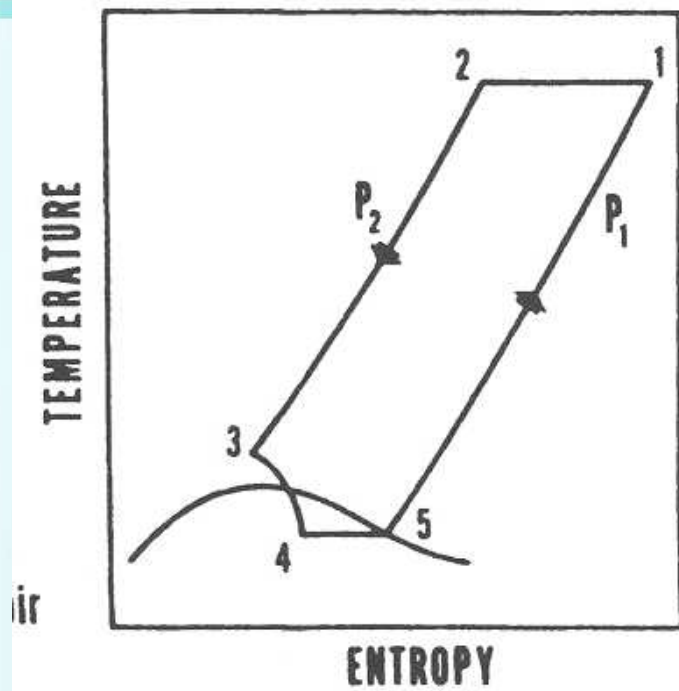
- What is cryogenics?
 - Nitrogen boils at $-320\text{ }^{\circ}\text{F}$
 - Argon boils at $-303\text{ }^{\circ}\text{F}$
 - Oxygen boils at $-297\text{ }^{\circ}\text{F}$
- Carl von Linde, 1985

General Theory

Linde Process

- Simplest liquefaction cycle
- Compressor, heat exchangers, J-T valve
- Valve Operation below inversion T and P

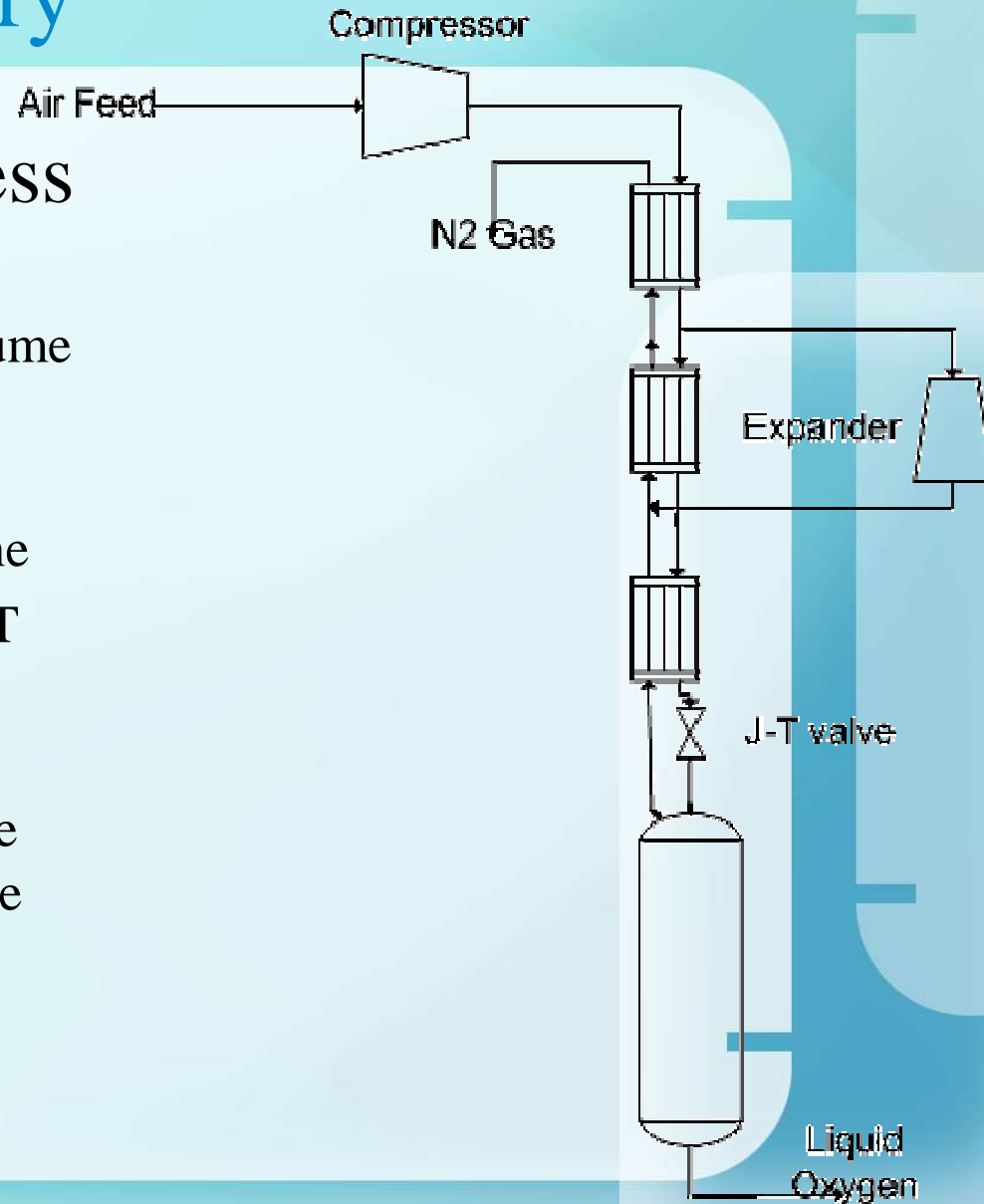


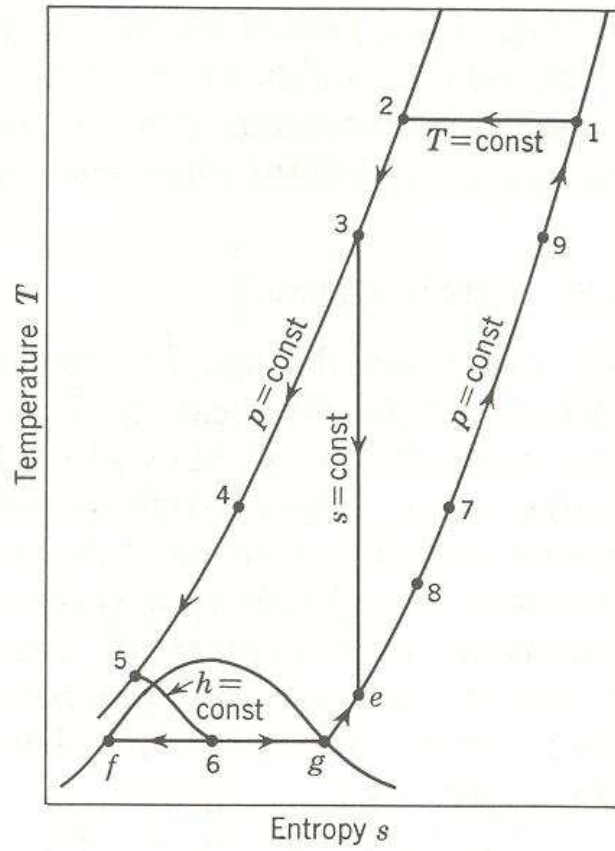


General Theory

Claude Process

- Modern high volume Cryo-plants
- Compressor, HX, Expansion Turbine
- Below inversion T and P spec. not required
- Hybrid of both the Brayton and Linde Cycle





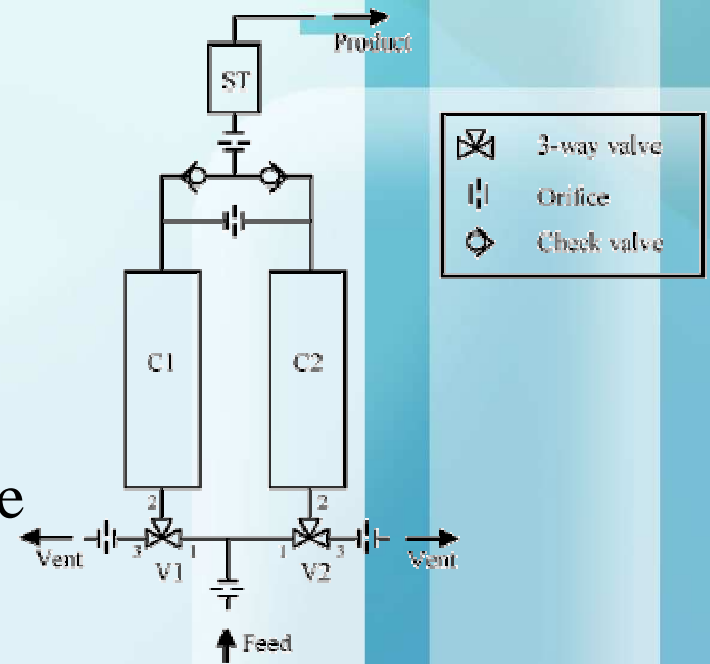
General Theory

Pressure Swing Adsorption

- A separation process through which a bed packed with molecular sieve or zeolite adsorbents are used to selectively adsorb a desired substance from a pressurized feed stream

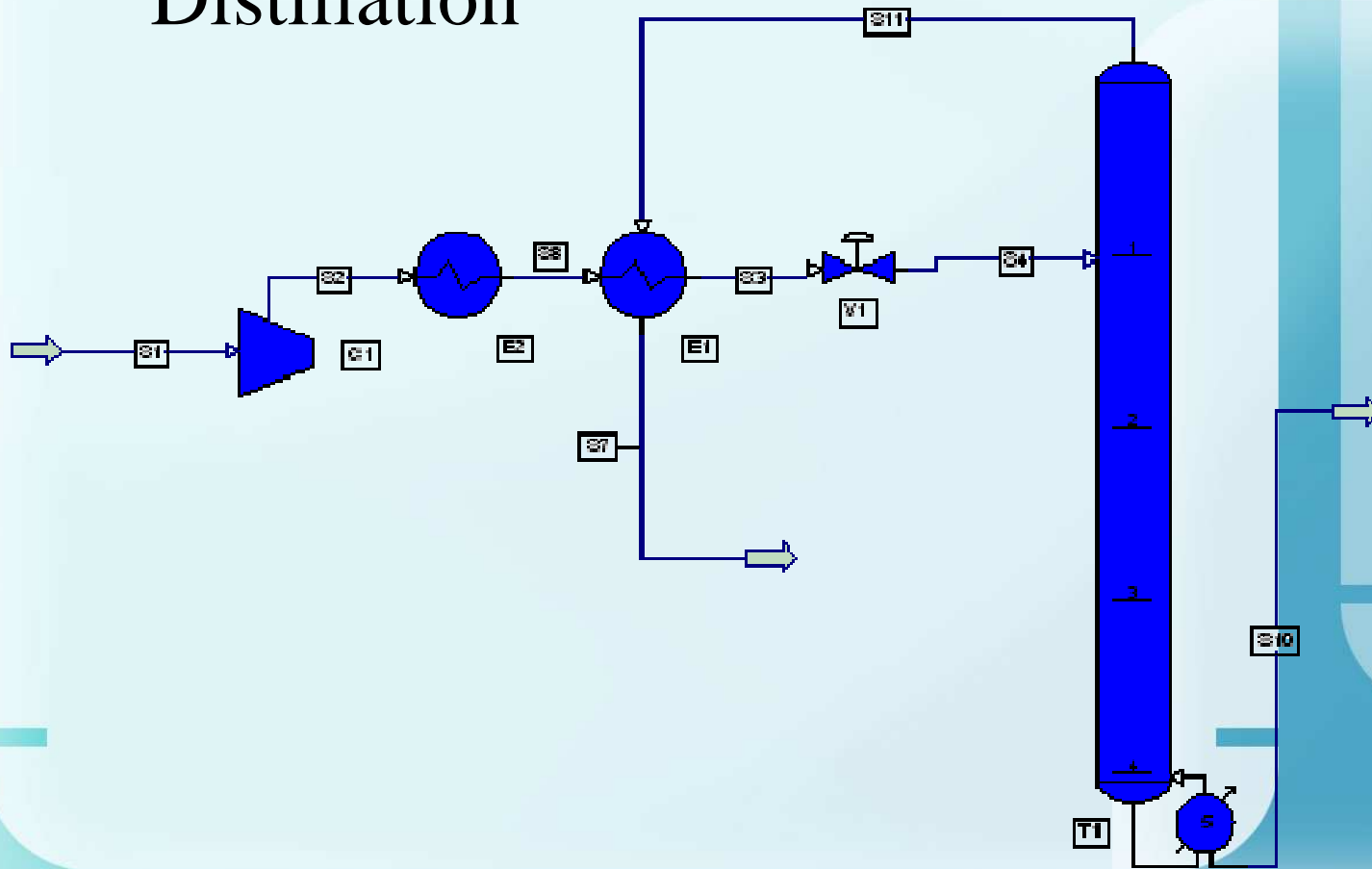
General Theory

- Two equal beds operate in alternating modes:
 - 1) adsorption
 - 2) desorption
 - this allows for continuous operation
- While one column is in mode 1 the other will always be in mode 2



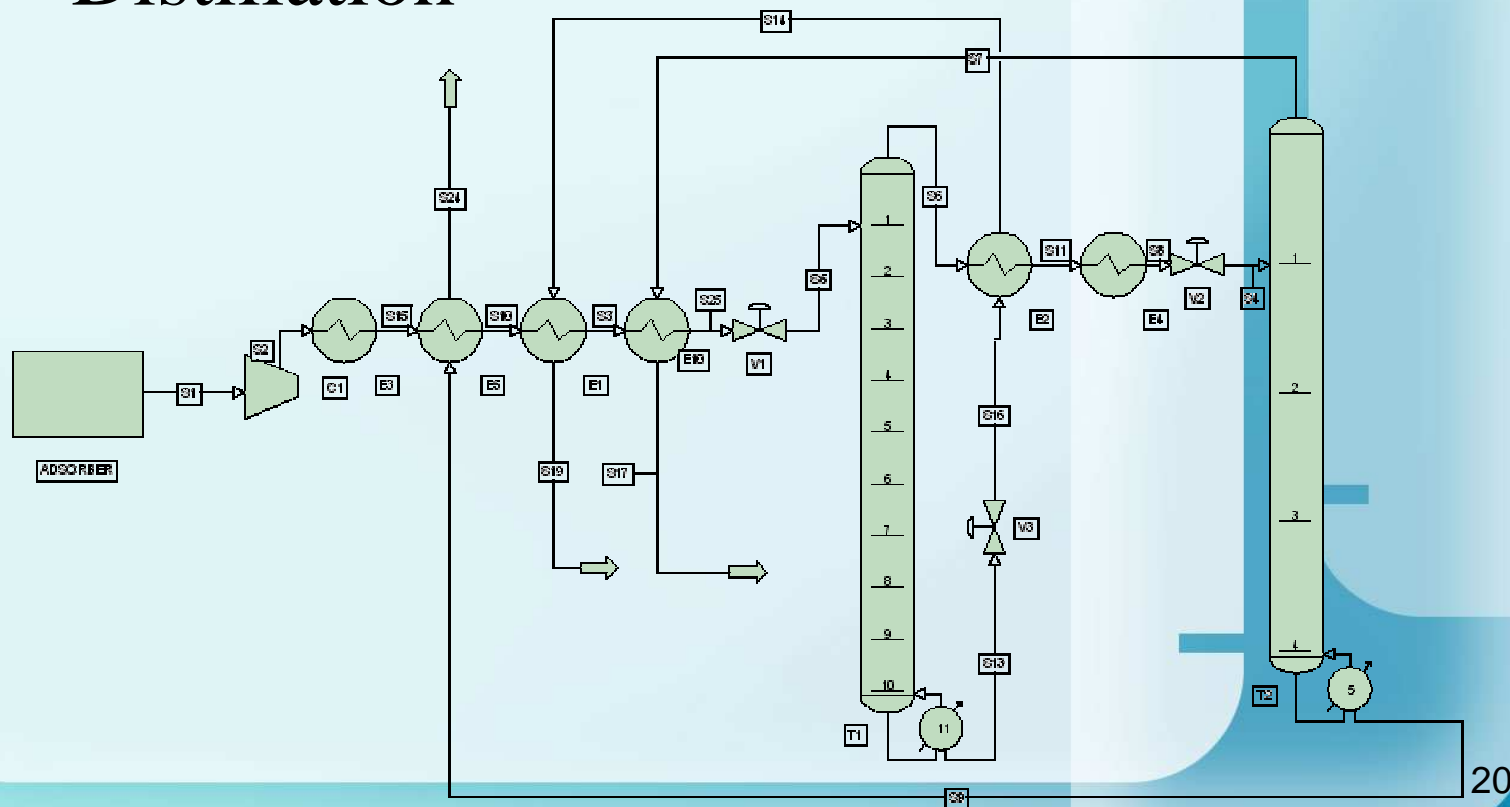
Three Options Considered

1. Air Feed into Cryogenic Distillation



Three Options Considered

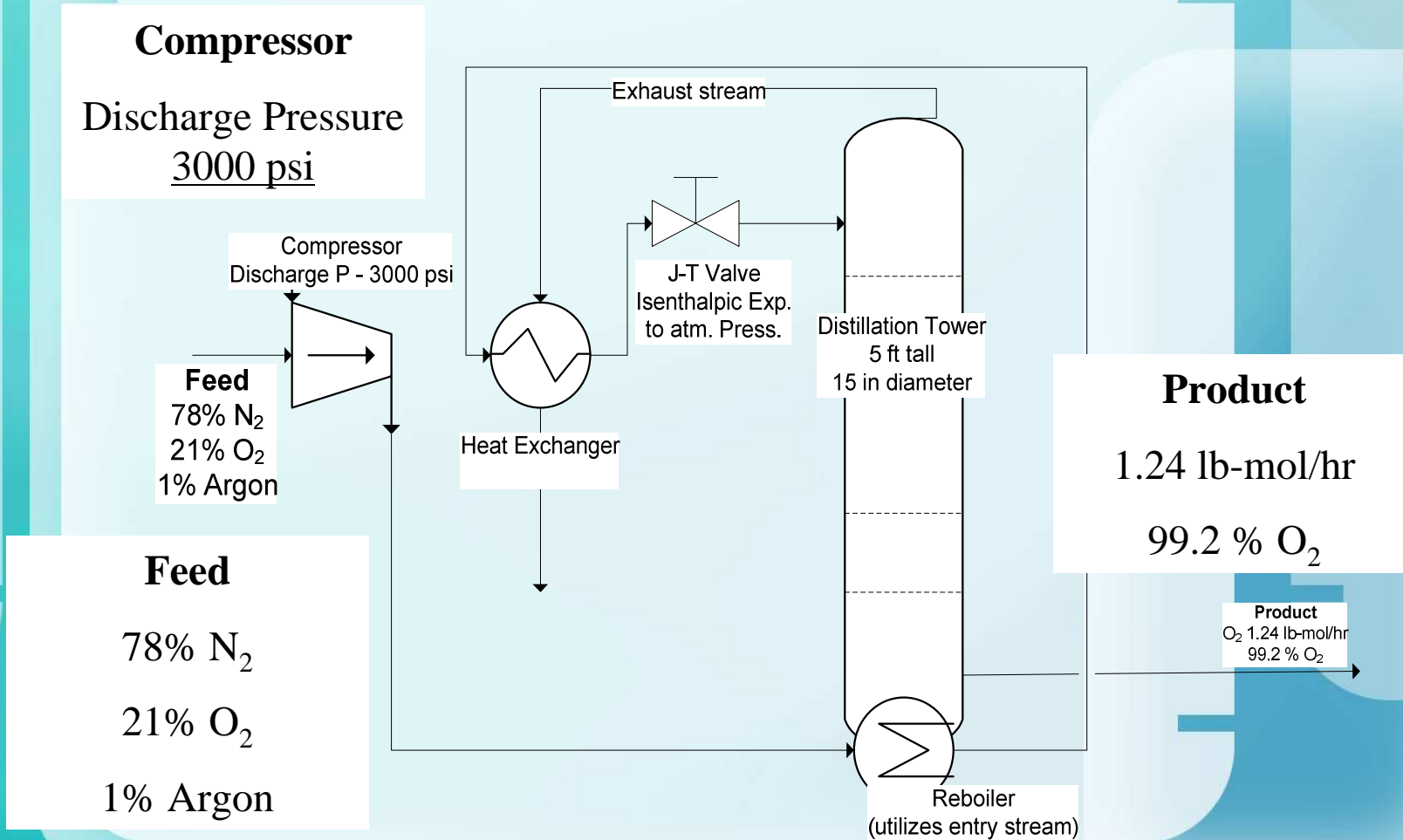
2. Air Feed into an N₂ Adsorber followed by a Cryogenic Distillation



Three Options Considered

3. Air Feed into an N₂ adsorber followed by Argon removal

Air Feed into Cryogenic Distillation



Air Feed into Cryogenic Distillation

- Required Flow Rate
(for 1.24 lb-mol/hr O_2)
95,000 ft³/hr
- Requires unfeasible energy to compress
Nearly 1,400 kW \rightarrow \$700,000/yr

Air Feed, N₂ Adsorber, Cryogenic Distillation

- Two Designs
 - With and Without Expander
- Results and conclusions

Air Feed, N₂ Adsorber, Cryogenic Distillation

- **First Design**

Expander

Expands to atmospheric

Product

1.24 lb-mol/hr

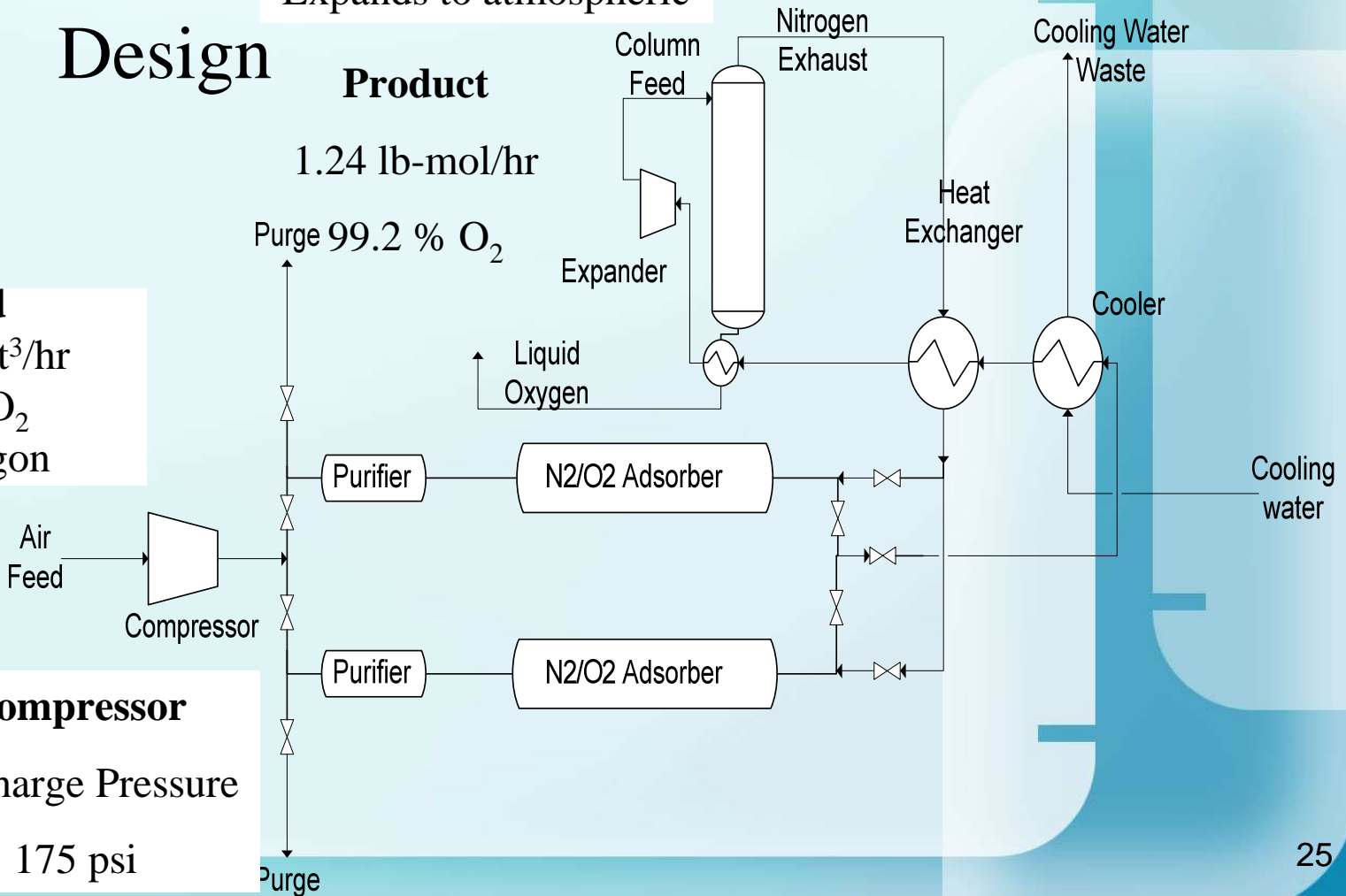
Purge 99.2 % O₂

Feed
21,000 ft³/hr
95% O₂
5% Argon

Compressor

Discharge Pressure

175 psi



Air Feed, N₂ Adsorber, Cryogenic Distillation

Column	\$5,300
Cold Box	\$34
<i>Compressor</i>	<i>\$105,000</i>
Heat Exchangers	\$14,560
<i>Expander</i>	<i>\$105,000</i>
Pressure Swing Adsorber - O ₂ /N ₂	\$3,530
Pressure Swing Adsorber - Purifier	\$1,900
Piping	\$1,900
Total Equipment Cost	\$237,000

Air Feed, N₂ Adsorber, Cryogenic Distillation

Operating Cost

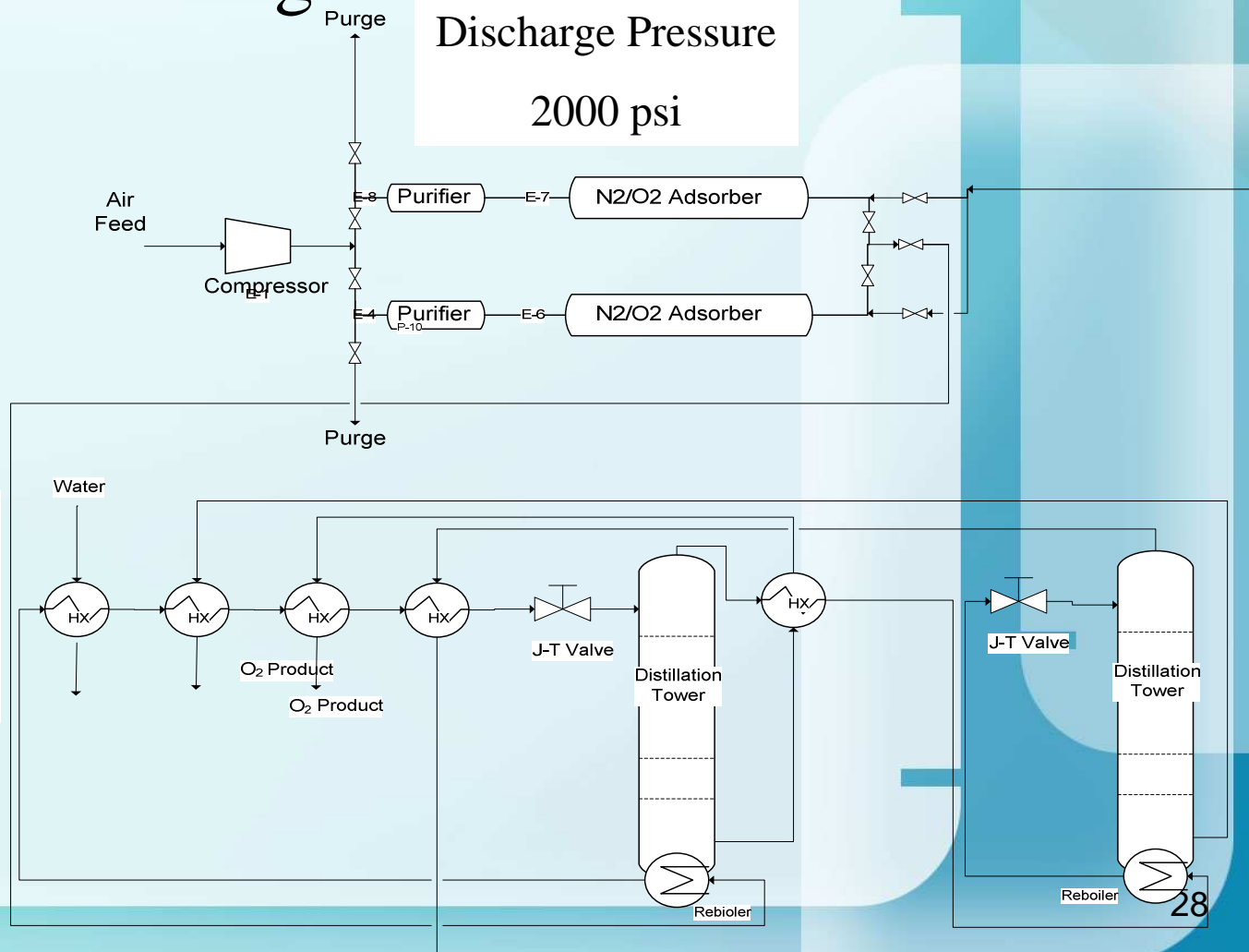
Compressor Power	\$35,300
Water	\$900
Total	\$36,200

Air Feed, N₂ Adsorber, Cryogenic Distillation

- Second Design

Compressor

Discharge Pressure
2000 psi

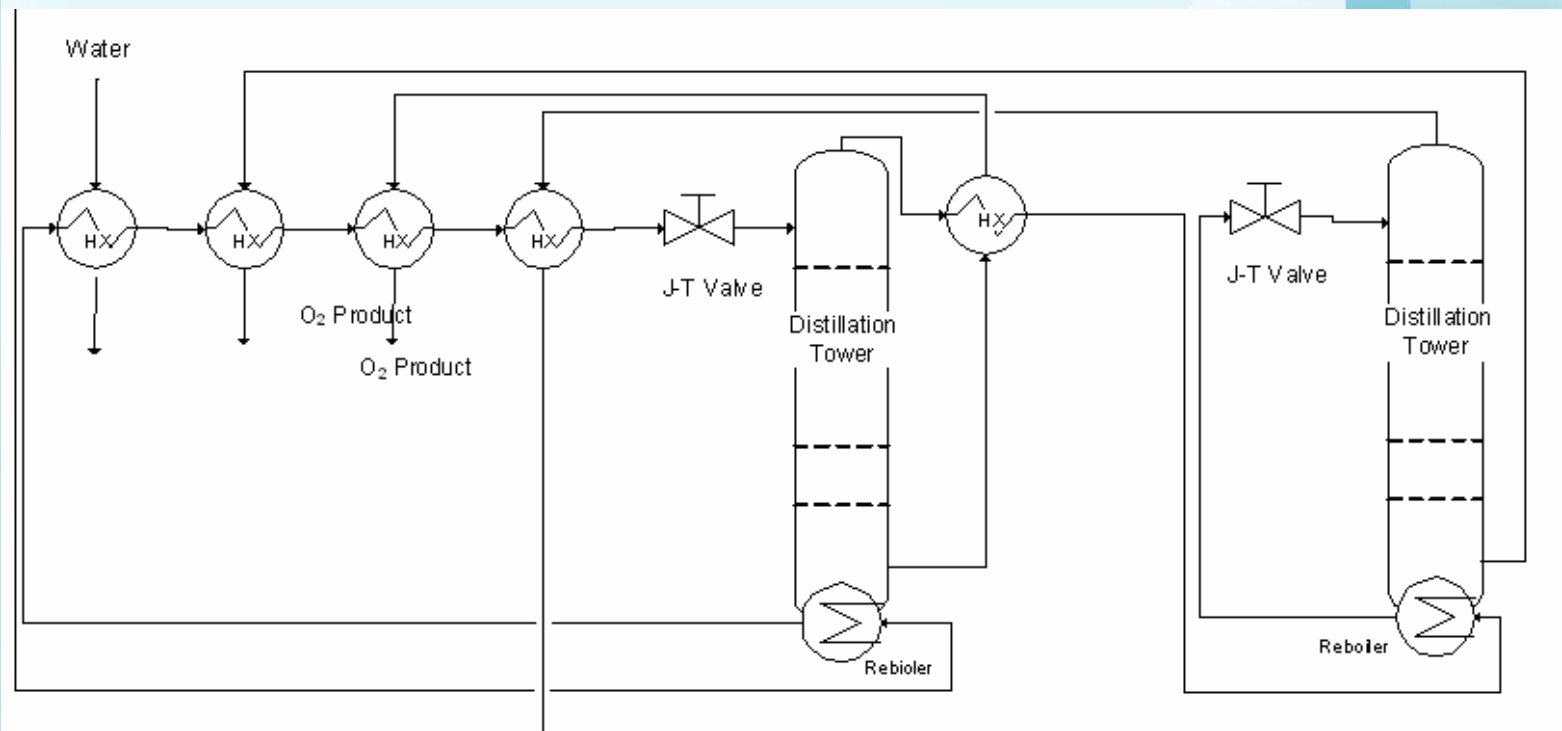


Feed (to cryo process)

8,000 ft³/hr

95% O₂

5% Argon



Air Feed, N₂ Adsorber, Cryogenic Distillation

Equipment Costs

Column:	\$10,600
Cold Box:	\$100
Compressor:	\$200,000*
Heat Exchanger:	\$4,000
Adsorber (O ₂ /N ₂):	\$3,500
Adsorber (Purifier):	\$2,000
Piping:	\$2,300
Total Capital Cost	\$222,500

* RIX Industries, Rick Turnquist Sales Engineer

Air Feed, N₂ Adsorber, Cryogenic Distillation

Total Operating Cost	\$/yr
Compressor Power (130 kW)	\$70,000
Water	\$5000
Total	\$75,000

- OG&E Electricity: \$0.058/kWh
- OKC Water: \$0.255/1000 ft³
 - 3000 ft³/hr

Cost Comparison

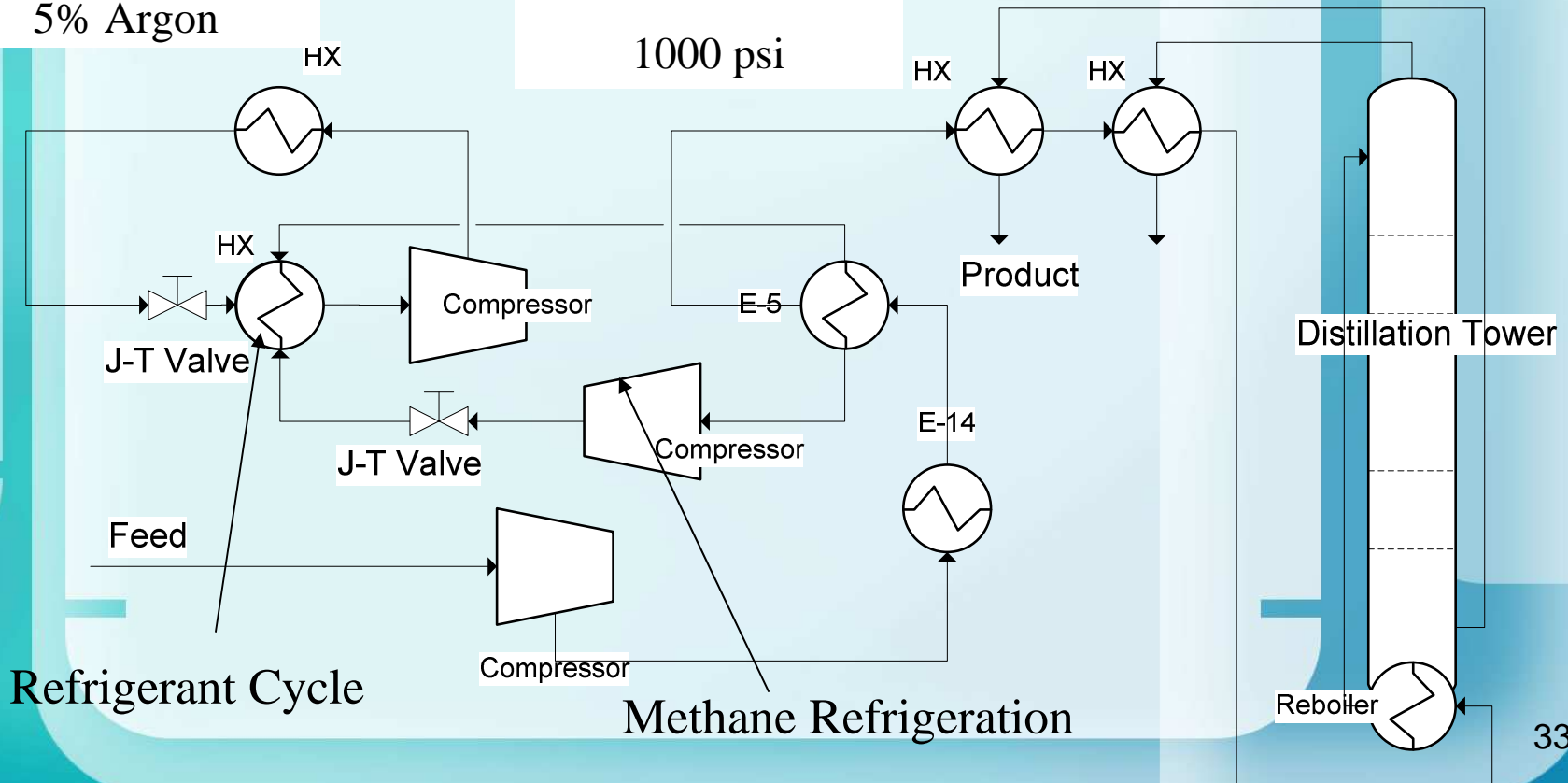
- Competitor
 - Delivered: *\$19,000 per year*
- Proposed first design
 - Total cost per year: \$60,000
 - Operating cost: \$36,000 per year
- Proposed second design
 - Total cost per year: \$97,250
 - Operating cost: \$75,000 per year

How Does a Plant Do It?

- Disregarding capital costs

Feed
1,800 ft³/hr
95% O₂
5% Argon

Compressor
Discharge Pressure
1000 psi



How Does a Plant Do It?

- 20 kW energy
- Results in only \$10,500/year energy costs
- Compared to \$19,000/yr distribution price

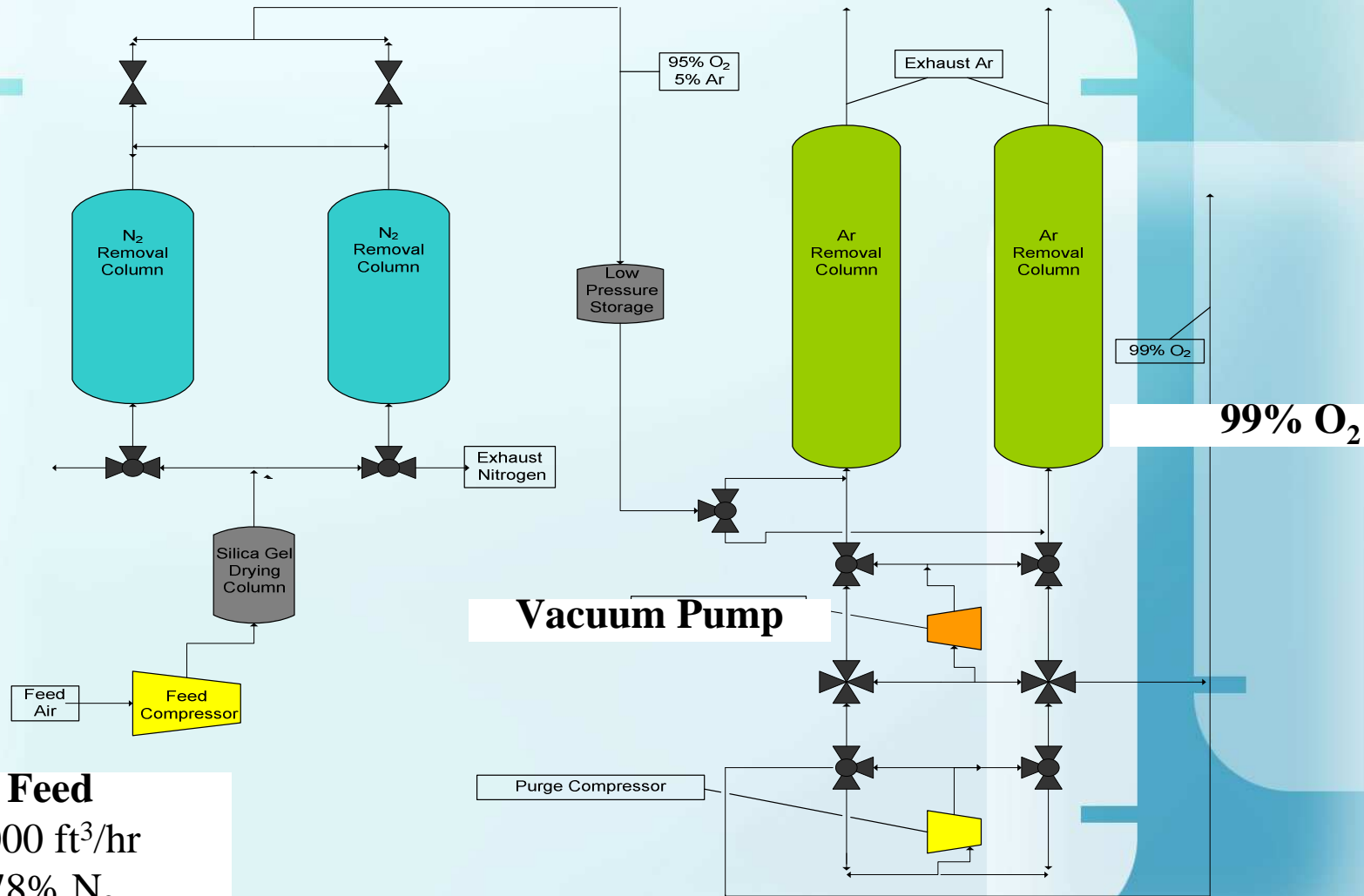
Cryogenic Distillation Conclusions

- The process is possible
- Energy costs are appeased by design incorporating more equipment
- Capital cost increase due to more equipment inhibits typical hospitals from making such large investments
 - Meaning → NO SAVINGS

Air Feed, N₂ Adsorber, Argon Adsorber

- Due to the infeasibility of the designed cryogenic system, a system utilizing Pressure swing adsorption to remove both N₂ and Argon removal was designed and examined

Air Feed, N₂ Adsorber, Argon Adsorber— PFD



Feed
4000 ft³/hr
78% N₂
21% O₂
1% Argon

Nitrogen Removal

Langmuir isotherm for multi-component adsorption

$$q_i = Q_{\max} \frac{b_i P_i}{1 + \sum_{j=1}^N b_j P_j}$$

- q_i = loading (mol/kg) on the adsorbent
- Q_{\max} = maximum loading (mol/kg) on the adsorbent
- N = the total number of components
- P_i = the partial pressure of component i
- Q_{\max} and b_i are given for adsorbent Oxysiv 5

Nitrogen Removal

$$Q_F c_F t_x = q_F M L_x / L_B$$

(Equilibrium driven: mass transfer effects negligible)

Q_F : volumetric feed flowrate

c_F : solute feed concentration

t_x : time of the front at position L_x

M : adsorbent mass in bed

L_x : distance traveled by the front

L_b : length of the bed

q_F : loading per mass of adsorbent

Nitrogen Removal

- Column specifications (per column)
 - Height: 7.2 ft
 - Column diameter: 1 ft
 - Column volume: 5.6 ft³
 - Adsorbent weight (Oxysiv 5): 109 kg (240 lbs.)

Argon Removal

Options

- Equilibrium PSA
- Rate based PSA

Argon Removal

- Equilibrium PSA
 - Operates similar to N₂ removal system
 - O₂ and Ar have similar physical properties and adsorption isotherms
 - Nearly equal amounts adsorbed resulting in lower yields

Argon Removal

Langmuir-Freundlich isotherms for O₂ and Ar

$$T = 30^\circ \text{ C.: } n_{Ar} = \frac{8.875C_{Ar}}{1 + 0.0041C_{Ar}} ; n_{O_2} = \frac{7.363C_{O_2}}{1 + 0.00307C_{O_2}}$$

$$T = 60^\circ \text{ C.: } n_{Ar} = \frac{5.222C_{Ar}}{1 + 0.0025C_{Ar}} ; n_{O_2} = \frac{4.155C_{O_2}}{1 + 0.00166C_{O_2}}$$

$$T = 90^\circ \text{ C.: } n_{Ar} = \frac{3.206C_{Ar}}{1 + 0.00117C_{Ar}} ; n_{O_2} = \frac{2.629C_{O_2}}{1 + 0.00108C_{O_2}}$$

Argon Removal

- Kinetic (rate based) separation
 - O₂ adsorbs at a much higher rate than Ar
 - Obtains 99% purity stream by the adsorption oxygen
 - BF-CMS (adsorbent) produces .01157 kg product/kg of adsorbent
 - 52.22% yield

Argon Removal

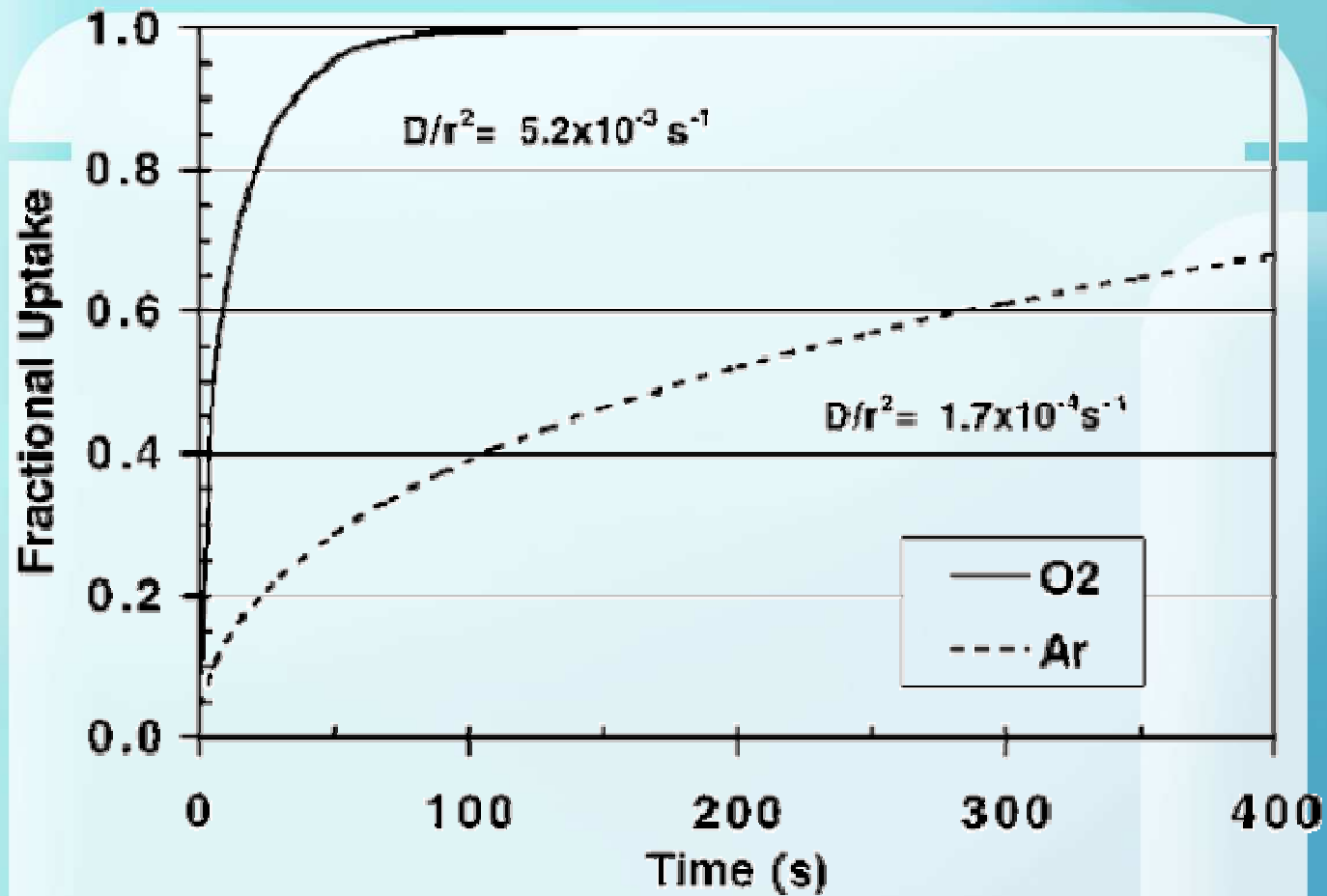
Rate based separation design equations

- Linear Driving Force Model

$$\frac{\partial \bar{q}}{\partial t} = \frac{15D_e}{R_p^2} \left(q_{R_p} - \bar{q} \right)$$

- t = time
- D_e = effective particle diffusivity
- R_p = radius of a particle
- q_{R_p} = loading at particle surface
- \bar{q} = average loading of component on adsorbent bed

Argon Removal



Fractional Uptake vs. time for Oxygen and Argon on Bergbau-Forschung CMS

Argon Removal

- Column specifications (per column)
 - Height: 16.4 ft
 - A Column diameter: 2.5ft
 - Column volume: 80.7 ft³
 - Adsorbent weight (BF-CMS): 1554 kg (3425.9 lbs)

Operating conditions

- Nitrogen system
 - Inlet flowrate of 4000 ft³/hr
 - Feed Air compression to 45 psia
 - Breakthrough time of 1 minute (cycle time of 2 min)
- Argon system
 - 1.24 lbmol/hr product flowrate
 - Air compressed to 2 atm
 - Desorption takes place at .2 atm
 - 99% product oxygen

Materials

	N ₂	Ar
Metal	ft ²	ft ²
Adsorption Columns (Al) \$1.5 / ft ²	23	260
Low pressure storage tank (Al)	12	
Dryer Canister (Al)	24	
Frame (Steel) \$2 / ft ²	25	265
Adsorbents	lb	lb
Oxysiv 5 adsorbent \$5.5 / lb	480	
BF-CMS \$3 / lb		6850
Silica gel \$2 / lb	52	

Equipment Cost Summary

	Price	N ₂	Ar
Piping	\$ / ft	ft	ft
1/2" Sch. 40 Copper	3.61	6	10
Compressors	\$	# of items	# of items
Feed	5365	1	
Purge	150		1
Other parts	\$/item		
Vacuum pump	100		1
Fan	5	2	2
3-way solenoid valve	86	2	2
Check valve	20	2	2
Control Computer	1000	1	

System Equipment Costs

Final Cost	\$31,600
<i>Additional Costs (based on need)</i>	
Tank fill Compressor	\$2,500.00
High pressure storage tank	\$150.00
Final Cost with Additions	\$34,200

Results

- System occupies 20 by 20 ft² area
- Yearly energy costs- \$8,500
- Average yearly cost of \$15,150 over 10 year life of machine

Comparison

- Average yearly distribution costs
\$19,000
- Average yearly O₂n-site generator costs
\$8,500
- Average Yearly savings
\$4,000

Business Plan for Adsorption System

- Open market for this type of equipment
 - Hospital need
 - Dependence on distributors
 - Stability of product price
- Oklahoma
 - Approximately 350 medical facilities

Business Costs

Fixed Costs

Tools	\$10,000
Truck	\$90,000
Trailer	\$17,500
Clerical Supplies	\$2,000
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Total	\$119,500

Operating Costs per year

Salaries	\$280,000
Insurance and Permits	\$23,900
Equipment Maintenance	\$11,950
Fuel	\$39,000
Total	\$354,850

Demand Model

- Factors include:
 - Convenience
 - Maintenance
 - Space
 - Reliability
 - Safety
- H is the product appeal determined on the demand factors

Demand Model

- β represents product preference
 - H_c is competitor appeal
 - H_d is new design appeal

$$\beta = \frac{H_c}{H_d} = .92$$

Factor	w_i	Y_d	y_d	Y_c	y_c
Convenience	0.40	9	0.90	7	0.70
Maint./Op. Cost.	0.20	6	0.60	7	0.70
Space	0.20	6	0.60	6	0.60
Reliability	0.10	9	0.90	8	0.80
Safety	0.10	7	0.70	8	0.80
Total=	1.00	$H_d = \sum w_i y_d =$	0.76	$H_c = \sum w_i y_c =$	0.70

Demand Model

- Consumer demand equation

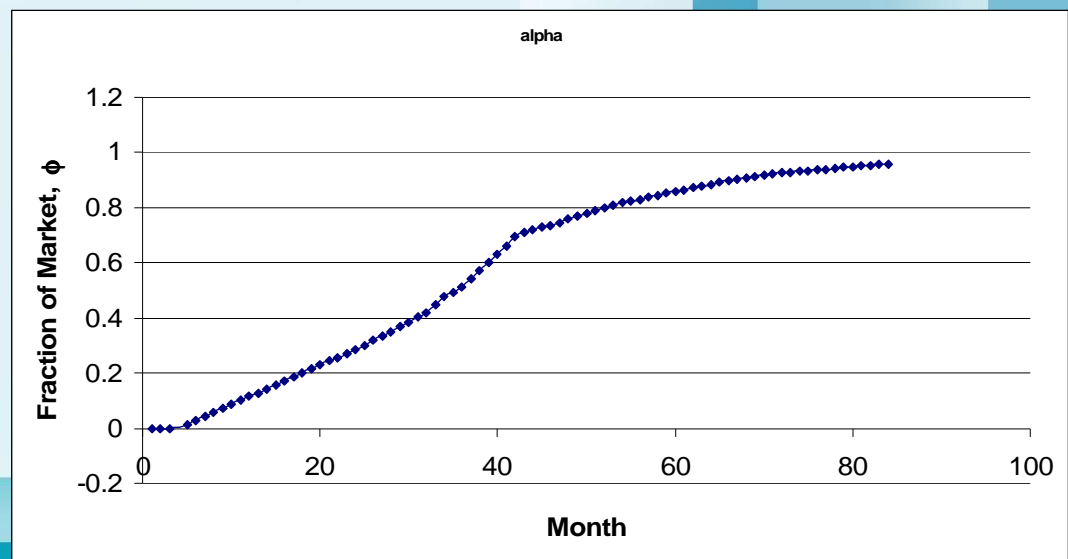
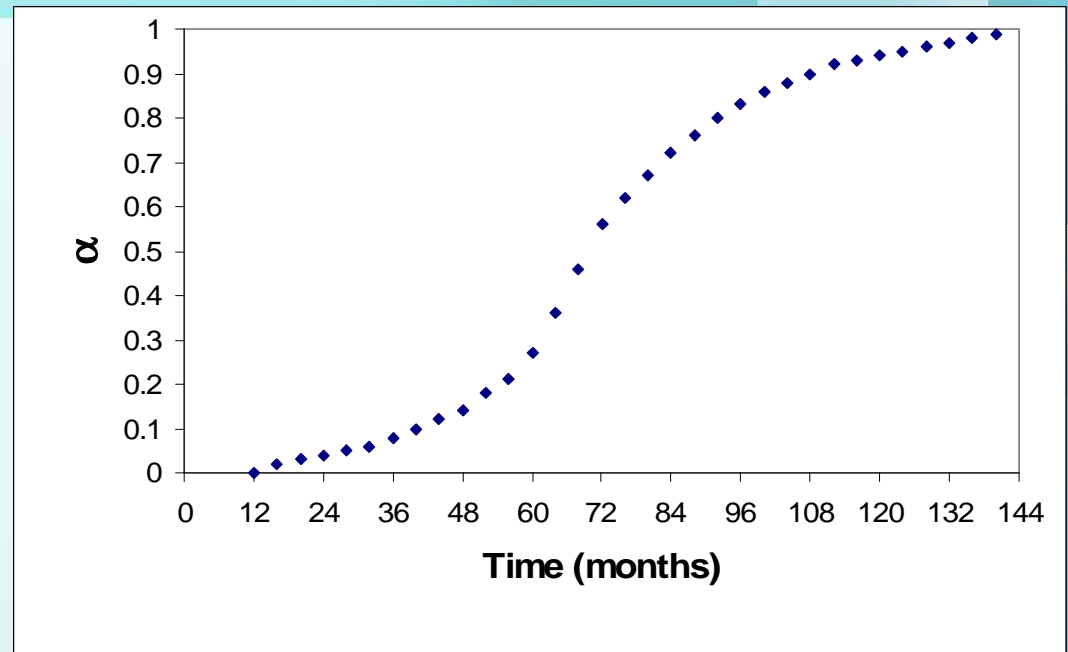
$$p_d d_d^\beta = \alpha p_c d_c \frac{d_d^\alpha}{d_c^\beta}$$

- Solve for new design demand, d_d

$$d_d = \frac{\alpha}{\beta} \frac{p_c}{p_d} (D - d_d)^{(1-\beta)} d_d^\alpha$$

Demand Based Sales Prediction

- α – The public knowledge of this product
- Varying Salespeople
- Responsibilities
 - Schedule meeting with potential clients
 - Repeat visits when requested or periodically
- Had to increase salespeople due to demand model behavior



Demand Based Sales Prediction

- Optimized selling price index
- Each selling price influenced demand of design
- Optimal selling price found
 - 1.9 times material costs
 - Upper limit for buy – gives yearly 20% savings
 - Gives total yearly customer cost of \$15,150

Economic Analysis

- Sale Factor greater than 1.9 cost of equipment
- Seven year NPV \$3.5 million
- Saturate market in
- Prediction
 - Total sale of 350 Systems

Conclusion

- Market does exist for on-site oxygen production
- On-site cryogenic oxygen production not feasible due to high capital costs
- *Adsorption system is economically feasible and is recommended*

Acknowledgements

- **Tom Reed**
- **Donovan Howell**

Questions